Code: C16 M-303

6244

BOARD DIPLOMA EXAMINATION MARCH/APRIL - 2019 DIPLOMA IN MECHANICAL ENGINEERING THERMAL ENGINEERING-I THIRD SEMESTER EXAMINATION

Time: 3 Hours Total Marks: 80 RUKRISHNA DISTA

PART - A $(3m \times 10 = 30m)$

Note 1:Answer all questions and each question carries 3 marks

2: Answers should be brief and straight to the point and shall not exceed 5 simple sentences

- 1. Write the expressions for C_p and C_v in terms of R and γ
- 2. Define Temperature
- 3. Define Heat Pump
- 4. Represent the following processes on T-S diagram
 - a. Constant pressure process. b. Isochoric proces
- 5. Represent throttling process on P-V and T-S diagram
- 6. What is the effect of compression ratio on efficiency of Otto Cycle?
- 7. Write the Dulong's formula to find HCV of the fuel
- 8. What are the specific requirements of fuel injection system in compression ignition engines?
- 9. 1. Find the heat energy to exhaust gas using the following

data

Temperature of Exhaust gas

 $=387.5^{\circ}c$

Temperature of engine room

 $=20^{\circ}c$

ir fuel ratio

=20

pecific heat of exhaust gas

 $=1.05 \text{ kJ/kg-}^{\circ}\text{K}$

el consumption

=0.128kg/min

10. Draw the theoretical indicator diagram for a single stage compressor

Code: C16 M-303

PART - B $(10m \times 5 = 50m)$

Note 1: Answer any five questions and each question carries 10 marks

2:The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer

- 12. A mass of an ideal gas has initial temperature of 16°C. If the temperature is raised to 280°C, find the change of internal and enthalpy. Assume specific heat at comb.

 kJ/kg°K and specific heat.
- 13. In a steady flow open system, a fluid substance flows at the rate of 10 kg/sec. It enters the system at a pressure of 600 kN/m², a velocity of 150 m/sec with internal energy 2000 kJ/kg and specific volume of 1 m³/kg. It leaves the system at a pressure of 125 kN/m², a velocity of 100 m/sec with internal energy 1600 kJ/kg and specific volume of 2 m³/kg. During its passage through the system, the substance has a lost by heat transfer of 40 kJ/kg to the surroundings. Determine the power of the system, stating whether it is from or to the system. Neglect any change of gravitational potential energy
- 14 2 kg of air at pressure of 875 kN/m² occupies a volume of 0.35 m³. The air is then expanded to a volume of 2 m³ at constant pressure. Find the work done and heat transfer during the process. Take R = 0.287kJ/kg K and $C_v = 0.717 kJ/kg K$
- 15. In an ideal Ott cycle, the air at the beginning of isentropic operation is 1 bar and 15 The ratio of compression is 8. The heat added is 10008 kJ/k during constant volume proess. Take $\gamma:1.4$ and C =0.714 kJ/kgK. Determine
 - . The maximum temperature in the cycle
 - b. The air standard efficiency
 - c. The work done per kg of air
 - d. The heat rejected per kg of air
- 16 Explain with the help of line diagram, the working principle of simple carburettor

17. The following observations were made during a trial on two stroke engine for half an hour when it was running at 250 rpm Stroke length=500 mm

Diameter = 200 mm

Mean effective pressure=6 bar

Reading of spring balances=1390 N and 90 N

Mean circumference of brake drum=3600 mm

Fuel consumed=5.4 kg

Calorific value of fuel=44000 kJ/kg

Determine a) Mechanical efficiency b) Indicated thermal efficiency & c) Brake thermal efficiency

^{18.} Air from an initial conditions of 25°C and 1 bar is compressed in two stages according to the law $pV^{1.25}$ = constant and with complete intercooling to a pressure of 36 bar. Estimate the minimum work required and intermediate pressure Cp =1.05 kJ/kg°K and R = 0. 29 kJ/kg°K -xxxminimum work required and intermediate pressure.