

со9-м-305

3249

BOARD DIPLOMA EXAMINATION, (C-09)

SEPTEMBER/OCTOBER - 2020

DME—THIRD SEMESTER EXAMINATION

THERMAL ENGINEERING-I

Time : 3 hours]

[Total Marks : 80

Uses of steam tables and Mollier diagram is allowed. **PART**—A 3×10=30

Instructions : (1) Answer all questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- **1.** State the second law of thermodynamics.
- 2. State Avogadro's law.
- 3. What is meant by isothermal process?
- 4. What is throttling process?
- 5. Mention any six types of solid fuels.
- **6.** Define lower calorific value of a fuel.
- 7. Define air standard efficiency of the cycle.

* /3249

[Contd...

- 8. What is wet steam?
- **9.** Find the enthalpy of steam at a pressure of 20 bar and dryness fraction 0.85 using steam tables.
- **10.** Define coefficient of performance of a refrigerated system.

PART—B

10×5=50

Instructions : (1) Answer any five questions.

- (2) Each question carries ten marks.
- (3) Answers should be comprehensive and the criterion for valuation is the content but not the length of the answer.
- **11.** (a) Explain briefly about the concept of flow work.
 - (b) A closed system executes a process during which 10 kJ of heat is supplied to the system. Find the change in internal energy under the following conditions :
 - (i) 5 kJ of work is done on the system
 - (ii) 2.5 kJ of work is done by the system
- **12.** Explain about the following processes in gases with the help of *P*-*V* and *T*-*S* diagrams :
 - (a) Constant volume process
 - (b) Constant pressure process
- **13.** 100 kJ of heat is added to 0.5 kg of air at constant volume whose initial temperature and pressure are 30 °C and 200 kPa. Take C_{12} 0.718 kJ/kg-K. Determine—
 - (a) final temperature;
 - (b) final pressure;
 - (c) change in entropy.

* /3249

[Contd...

- **14.** Explain about the construction and working of bomb calorimeter with a neat sketch.
- **15.** Explain the working of Otto cycle with the help of *P*-*V* and *T*-*S* diagrams.
- **16.** (a) Derive the relationship between specific heats and gas constant.
 - (b) A Carnot cycle operates between the temperature limits of 300 °C and 40 °C. The heat supplied to the system is 120 kJ. Determine—
 - (i) air standard efficiency;
 - (ii) work transfer.
- 17. 1 kg of steam at a pressure of 10 bar absolute and 0.8 dryness fraction expands during a non-flow polytropic process according to the law PV^{13} constant, until its pressure becomes 2.8 bar. Determine
 - (a) final condition of steam;
 - (b) work done;
 - (c) change in internal energy;
 - (d) heat transfer.
- 18. Write short notes on the following :
 - (a) Open-cycle air refrigeration
 - (b) Closed-cycle air refrigeration

* * *