

C16-EE/CHPP-104

6037

BOARD DIPLOMA EXAMINATION, (C-16) MARCH/APRIL—2017

DEEE—FIRST YEAR EXAMINATION

ENGINEERING CHÉMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours] [Total Marks: 80

PART—A

 $3 \times 10 = 30$

Instructions: (1) inswer all questions.

(2) Each question carries **three** marks.

- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- 1. State Pauli's principle with example.
- State three properties of ionic compound.
- **3.** Define the following:
 - (a) Mole
 - (b) Normality
 - (c) Solution
- **4.** What are buffer solutions? Mention two uses.
- **5.** Define the following :
 - (a) Electrode potential
 - (b) EMF of the cell
- **6.** What are the causes of hardness of water?

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7.	Exp	plain addition polymerization with an example.				
8.	Me	ention the composition and uses of (a) water gas (b) producer s.				
9.	Wr	ite any three causes of water pollution.				
10.	0. Define the following terms :					
	(a)	Pollutant				
	(b)	Biodiversity				
	(c)	Ecosystem				
		PAR 10 'B 10×5=5	50			
Inst	ruct	tions: (1) Answer any five questions.				
		(2) Each question carries ten marks.				
		(3) Answers should be comprehensive and the criterion for valuation is the content but not the length the answer.				
11.	(a)	Write the postulates of Bohr's atomic theory with two limitations.	7			
		Calculate the oxidation number of Mn in $KMnO_4$ and Cr in $K_2Cr_2O_7$.	3			
12.5	(a)	Explain the Bronsted-Lowry theory of acids and bases with examples.	5			
	(b)	Define molarity. Calculate the molarity of a solution if 5.3 gms of Na_2CO_3 is present in 250 ml of solution.	5			
13.	(a)	Write the composition and uses of German silver and brass.	4			
	(b)	State the differences between metals and non-metals.	6			
14.	(a)	Explain Faraday's laws of electrolysis.	5			
	(b)	What is electro chemical series? Mention its significance.	5			
15.	, ,	Explain the formation of a composition cell and stress cell.	5			
	. ,	State and explain the five factors which influence the rate of corrosion.	5			

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16.	(a)	Write the essential qualities of drinking water.	4
	(b)	Describe the permutit process of softening of hard water.	6
17.	(a)	Mention the preparation and uses of (i) Buna-S and (ii) Neoprene rubber.	5
		setting plastics.	5
18.	(a)	Define 'air pollution'. Explain the causes of air pollution.	7
	(b)	Define the following:	3
		(i) Producers	
		(ii) Consumers	
		(iii) Decomposers	
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	17.	(b) 17. (a) (b)	(ii) Neoprene rubber.(b) Write the differences between thermo-plastics and thermo-

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