

6019

BOARD DIPLOMA EXAMINATION, (C-16) OCT/NOV-2018 DCE-FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours [Total Marks: 80

PART—A

3×10=30

Instructions:

- (1) Answer **all** questions.
- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- **1.** Define oxidation number. Calculate oxidation number of Mn in $KMnO_4$.
- 2. Define unit cell and co-ordination number.
- **3.** How many grams of Na₂CO₃ present in 5 moles of it.
- **4.** Define butter solution. Give one example for each type of buffer.
- **5.** Write any three differences between metallic conductors and electrolytic conductors.
- 6. Define soft water and hard water.
- 7. Define plastic Give two examples for it.
- **8.** Define fuel. Write compsition and uses of coal gas.
- **9.** Define pollutant and contaminant.
- **10.** What is biodiversity. Mention any two threats to it.

PART-B $10 \times 5 = 50$

- **Instructions:** (1) Answer any **five** questions.
 - (2) Each questions carries **ten** marks.
 - (3) Answers should be comprehensive and the criteria for valuation are the content but not the length of the answer.
- **11.** a) Explain four quantum numbers.
 - b) State pauli's exclusion principle.
- **12.** a) Define normality . 40gms of sodium hydroxide is present in 100ml of solution. Find the normality (molecular weight of Na0H=40) of the solution.
 - b) Explain Lewis acid-base theory.
- **13.** a) Explain forth flotation process for concentration of sulphide ore.
 - b) Write any five characteristics of metals.
- 14. a) Explain Faradays Laws of electrolysis
 - b) Define a Galvanic cell. Explain construction and functioning of a Galvanic cell.
- a) Explain mechanism of rusting of iron.
 - b) Explain composition and stress cell with examples.
- a) Explain the ion-exchange process for softening of hard water.
 - b) Write any four essential qualities of drinking water.
- a) Define and explain Vulcanization process of natural rubber with necessary equations.
 - b) Write the differences between addition and condensation polymerization.

- **18.** a) Write the controlling method of water pollution.
 - b) What are primary and secondary pollutants? Give exampes.

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