

# C20-A-AA-AEI-CHST-BM-TT-MET-MNG-AM-WD-AI-AG-104

# 7004

# **BOARD DIPLOMA EXAMINATION, (C-20)**

## JUNE/JULY—2022

## FIRST YEAR (COMMON) EXAMINATION

#### ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours [ Total Marks: 80

#### PART—A

 $3 \times 10 = 30$ 

**Instructions:** (1) Answer **all** questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.
- **1.** Define unit cell. Write the structure and coordination number of NaCl, CsCl.
- 2. Define normality. Write mathematical formula and units for it.
- **3.** What is ionic product of water? Write value of ionic product of water at 25 °C.
- 4. What is salt bridge? How is it prepared? What is its use?
- 5. Define hard water. Write causes for temporary hardness.
- **6.** Write the method of preparation and uses of Teflon.
- 7. Define the fuel. Write the composition and uses of natural gas.
- **8.** Write the health aspects of usage of soft drinks.
- 9. Write three causes for air pollution.
- 10. Define pollutants, receptor and dissolved oxygen.

**PART—B** 8×5=40

**Instructions:** (1) Answer **all** questions.

- (2) Each question carries eight marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **11.** (a) Write the differences between orbit and an orbital and draw the shapes of s, p, d orbitals.

(OR)

- (b) State and explain Aufbau principle and Hund's rule.
- **12.** (a) Define solution. Classify solutions based on physical state with examples.

(OR)

- (b) Define pH. Calculate the pH of a solution containing 4 gm of NaOH in a liter of solution.
- **13.** (a) Explain the purification of copper by electrolytic refining.

(OR)

- (b) State Faraday's laws of electrolysis.
- **14.** (a) Explain composition concentration and stress cells with neat diagrams.

(OR)

- (b) Write short notes on (i) chlorination and (ii) coagulation.
- **15.** (a) Define vulcanization of rubber. Explain with chemical equation.

(OR)

(b) Explain greenhouse effect and acid rain.

**PART—C**  $10 \times 1 = 10$ 

**Instructions:** (1) Answer the following question.

- (2) The question carries ten marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **16.** Define polymerization reaction. Describe about addition and condensation polymerization reactions with suitable examples.

