

## C16-C-104/C16-CM-104

## 6019

## BOARD DIPLOMA EXAMINATION, (C-16) OCT/NOV-2017

## DCE—FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours ]

[ Total Marks : 80

PART—A

 $3 \times 10 = 30$ 

Instructions: (1) Answer all questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- 1. Write any three differences between orbit and orbital.
- 2. Give the number of electrons, protons and neutrons present in Na.
- **3.** Define mole. Calculate the number of moles present in 90 gm of water.
- 4. Define buffer solution. State any two uses of buffer solution.
- **5.** Write any three differences between electrolytic cell and galvanic cell.
- **6.** What are the disadvantages of using hard water?
- **7.** What are the elastomers? Give example.
- **8.** What are the characteristics of good fuel?

10.	De	fine air pollution. Write any two causes of air pollution.	
		<b>PART—B</b> 10×5=	:50
Ins	truc	tions: (1) Answer any five questions.	2
		(2) Each question carries <b>ten</b> marks.	۶
		(3) Answers should be comprehensive and the criteri for valuation is the content but not the length the answer.	
11.	(a)	What are the important postulates of Bohr's theory? Mention the limitation of this theory.	6
	(b)	Mention the limitation of this theory.  Draw the shapes of $s, p, d$ orbitals of acid and base with one	4
12.	(a)	Define equivalent weight of cacid and base with one example.	5
	(b)	Explain the concept of Lewis acid and Lewis base with examples.	5
13.	(a)	Define the following terms:  (i) Mineral  (ii) Ore	5
		(iii) Metallurgy	
	P.	(v) Flux	
	(b)	Explain calcination and roasting with examples.	5
14.	(a)	Explain the construction and functioning of a galvanic cell.	5
	(b)	A current of 2 amp passing through silver nitrate solution for 10 minutes deposits 1·4292 gm of silver. What is the electrochemical equivalent of silver?	5
* /60	19	2 [Conta	d

**9.** Write a short note on greenhouse effect.

(a)	What is the rust? Explain the mechanism of rusting of iron.	5
		5
(a)	Describe the method of municipal treatment of water.	6
(b)	Define osmosis and reverse osmosis.	4
(a)	Define and explain addition polymerization and condensation polymerization with examples.	5
(b)	Distinguish between thermoplastics and thermosetting plastics.	5
(a)	Explain about ecosystem, producers, consumers and decomposers.	5
(b)	Write the effects of water pollution.	5
	*** CETIL	
	(b)	<ul> <li>(a) Describe the method of municipal treatment of water.</li> <li>(b) Define osmosis and reverse osmosis.</li> <li>(a) Define and explain addition polymerization and condensation polymerization with examples.</li> <li>(b) Distinguish between thermoplastics and thermosetting plastics.</li> </ul>

\* **/6019** 3 AA7(A)—PDF