

C14-A-104/C14-AA-104/C14-BM-104/ C14-CHST-104/C14-AEI-104/C14-MET-104/

$c_{14-MNG-104/c_{14-IT-104/c_{14-TT}}}$

4004

BOARD DIPLOMA EXAMINATION, (C-14)

MARCH/APRIL-2016

FIRST YEAR (COMMON) EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time : 3 hours]

[Total Marks : 80

PART—A

3×10=30

Instructions : (1) Answer all questions.

- (2) Each question carries **three** marks.
- (3) Answers should be brief and straight to the point and shall not exceed *five* simple sentences.
- 1. State the charge and mass of fundamental particles.
- **2.** Define 'oxidation number'. What is the oxidation number of P in H_3PO_4 ?
- **3.** Define the terms 'solution', 'solute' and 'solvent'.
- **4.** What is the pH of $0.02 \ M H_2SO_4$ solution?
- 5. Define (a) electrode potential and (b) EMF.
- **6.** State any three disadvantages of using hard water in industries.

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- 7. Define (a) plastic and (b) elastomer. Give examples.
- **8.** Write the composition and uses of *(a)* water gas and *(b)* producer gas.
- 9. Define primary and secondary pollutants. Give examples.
- **10.** Define (a) producers and (b) consumers with examples.

PART—B

10×5=50

Instructions : (1) Answer any five questions.

- (2) Each question carries ten marks.
- (3) Answers should be comprehensive and the criterion for valuation is the content but not the length of the answer.

11.	(a)	Write any five differences between orbit and orbital.	5
	(b)	Define ionic bond. Explain the formation of ionic bond in MgO.	5
12.	(a)	Explain the Lewis theory of acids and bases.	5
	(b)	Define 'buffer solution'. State three applications of buffer solutions.	5
13.	(a)	Define the following terms :	4
		(i) Flux	
		(ii) Gangue	
		(iii) Ore	
		(iv) Slag	
	(b)	List any six characteristics of metals.	6
14.	(a)	Write the differences between metallic and electrolytic conduction.	4
	(b)	State the first law of Faraday. A current of 0.5 ampere is passed through CuSO ₄ solution for 45 minutes. Calculate the weight of copper deposited. (Atomic weight	
		of copper 63 5)	6

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15.	(a)	Describe the impressed voltage method.	5
	(b)	Explain five factors which influence the rate of corrosion.	5
16 .	(a)	Mention any five essential qualities of drinking water.	5
	(b)	Define 'reverse osmosis'. List any three advantages of reverse osmosis.	5
17.	(a)	Mention the advantages of plastics over traditional materials.	5
	(b)	Write the differences between thermoplastics and thermosetting plastics.	5
18.	(a)	Define the following with examples :	4
		(i) Nonrenewable energy sources	
		(ii) Renewable energy sources	
	(b)	Define 'biodiversity'. Briefly discuss any four threats to biodiversity.	

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